Assessment of Sirwan River Water Quality from Downstream of Darbandikhan Dam to Kalar District, Kurdistan Region, Iraq

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Abstract

This study was performed to evaluate the Siwran River water quality between Darbandikhan Downstream Dam and Kalar city for domestic and irrigation uses. Seven stations from different sites have been selected along the Sirwan River, and four replications from each station were taken. The parameters of water quality used in this work are (Turbidity, pH, Total hardness, Magnesium, Calcium, Sulphate, Nitrate, Chloride, Conductivity μs/cm, TDS). Data analysis shows that the water quality parameters of Sirwan River are not compatible with the drinking water standards especially the concentrations of Aluminum and Iron which show increasing levels than the maximum allowable levels for drinking water standards. In addition, to classifying water quality and evaluation its suitability for irrigation purposes, SAR, RSC, and ESP were calculated following standard equations and found experimentally as 0.5, 1.7, and 5.3 respectively. The results of the study revealed that the Sirwan River water should be used with good irrigation management techniques and soil salinity monitored by laboratory.

Keywords: Water Quality, Sirwan River, Aluminum concentrations, Sodium Adsorption Ratio (SAR), Residual Sodium Carbonate (RSC), Exchangeable Sodium Percentage (ESP).

1. Introduction

Water resources play an important role on population growth in any living area. With increasing inhabitants, water demand growths; meanwhile, the world is facing with severe water crisis. Rivers are most vital resources of fresh water in the world [1]. The water of rivers must be managed in an appropriate way particularly for those rivers pass
through numerous countries [2]. Sirwan River is the source of lifeline to almost one million residents of Kurdistan Iraq. The cities along Rirwan River depend mostly on it for domestic, municipal, agriculture and other purposes. Water quality mainly depends on the physical, chemical and biological properties. These characteristics give an indicator on water use for a specific purpose. Fresher water is important to human health, agriculture, and environments. The quality of water is a critical issue in the world. It is clear that the quantity of chemical composition in water is changed as a result of the changing the quantity of surface and ground water in a specific area [3]. For this reason, monitoring physiochemical properties of water are essential issue to deal with. The appropriate properties of water quality should include the measuring of pH, temperature, dissolved oxygen, essential and toxic cation elements, anions, electrical conductivity, total dissolved solid, chemical oxygen demand, biochemical oxygen demand, total organic carbon, taste, color, and extra [4]. This study has been mainly conducted in order to measure and analyze the water quality parameters of Sirwan River such as electric conductivity EC, total dissolved solids TDS, sodium adsorption ratio SAR, magnesium hazard MH, pH, residual sodium carbonate RSC, that could potentially impact on the quality of water for drinking and irrigation crops.

2. Methodology

2.1 Study Area
The study area consists of seven sites along the river were selected between Darbandikhan downstream dam and Kalar District. Four replications sample from each site were selected. These sites are very important for drinking water as the study area was dense of the population along the banks of the river, as well as the presence of some industrial activities. Also, it includes a number of fallings the wastewater, which are distributed on both sides of the river. Therefore, dangers of various biological, chemical and physical pollutants could be existed, which can affect the quality of the river water as a source of water for processing drinking and irrigation water.

Figure (1) shows the location of sampling stations from downstream of Derbendixan River in Derbendixan area to Shexlenger village at down of Kalar district. Seven stations were selected along Sirwan River which starting from downstream of Darbandikhan Dam, Maydan, Bawanur, Isayi, Qulasutaw, Kalar, and Shekhlanger in south of the Kalar City. Four replication water samples were collected from each of these stations during April 2018 in 1-L Poly ethylene bottles that are rinsed several times before filling.
2.2 Physicochemical Study

The samples were kept refrigerated and analyzed within 48 hours after collection in the laboratory of Chemistry Science Department in Garmian University. Various tests were conducted according to the Standard Methods for examination of water [5]. Some physicochemical parameters were studied in situ and composited at laboratory as required. The pH, EC and DO were measured immediately in situ by using a portable WTW conductometer and pH meter. TDS was measured using an HANNA instrument EC/TDS meter in a laboratory after the samples were arrived within 24 hours. (SO$_4^{2-}$), chloride (Cl$^-$), and (NO$_3^-$) ion concentrations were measured by means of SENTEK ion selective electrodes after 24 hours from sample collection. Some essential and toxic elements were analyzed using induced coupled plasma optical emission spectroscopy (ICPOES)

![Figure 1: Location of the sampling stations.](image-url)
For analysis of water samples by ICPOES, the samples were acidified by 2% HNO$_3$, and then left for 24 hours before analysis. These parameters mainly consist of certain physical and chemical characteristics of water that are used in the evaluation of agricultural water quality.

3. Results and Discussion

Physicochemical properties were measured and their results can be seen in tables 1, 2, 3 and 4 respectively.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Turbidity (NUT)</th>
<th>pH</th>
<th>EC (μS/cm)</th>
<th>Total Hardness (mg/l)</th>
<th>TDS (mg/l)</th>
<th>D.O (mg/l)</th>
<th>NO$_3^-$ (mg/l)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Darbandixan</td>
<td>1.47</td>
<td>7.80</td>
<td>371</td>
<td>134.74</td>
<td>448.0</td>
<td>7.71</td>
<td>1.51</td>
</tr>
<tr>
<td>Maydan</td>
<td><strong>5.40</strong></td>
<td>7.60</td>
<td>455</td>
<td>137.02</td>
<td>456.0</td>
<td>7.73</td>
<td>4.56</td>
</tr>
<tr>
<td>Bawanur</td>
<td>2.80</td>
<td>7.63</td>
<td>481</td>
<td>141.39</td>
<td>486.0</td>
<td>7.13</td>
<td>3.34</td>
</tr>
<tr>
<td>Isayi</td>
<td>2.99</td>
<td>7.88</td>
<td>500</td>
<td>141.42</td>
<td><strong>611.0</strong></td>
<td>7.92</td>
<td>2.62</td>
</tr>
<tr>
<td>Qulasutyaw</td>
<td>2.88</td>
<td>7.60</td>
<td>485</td>
<td>140.52</td>
<td>485.0</td>
<td>7.45</td>
<td>3.66</td>
</tr>
<tr>
<td>Kalar</td>
<td>1.63</td>
<td>7.76</td>
<td>468</td>
<td>135.98</td>
<td>463.0</td>
<td>7.99</td>
<td>2.96</td>
</tr>
<tr>
<td>Shexlangar</td>
<td>1.72</td>
<td>7.35</td>
<td>788</td>
<td>143.82</td>
<td><strong>657.0</strong></td>
<td>6.70</td>
<td>3.37</td>
</tr>
<tr>
<td>Ave.</td>
<td><strong>2.70</strong></td>
<td><strong>7.66</strong></td>
<td><strong>506.86</strong></td>
<td><strong>139.27</strong></td>
<td><strong>515.1</strong></td>
<td><strong>7.52</strong></td>
<td><strong>3.15</strong></td>
</tr>
<tr>
<td>WHO</td>
<td>5.00</td>
<td>6.5 - 8.5</td>
<td>400-800</td>
<td>500.00</td>
<td>500.0</td>
<td>5.00</td>
<td>45.00</td>
</tr>
</tbody>
</table>

3.1 pH of Water

The (pH) plays important roles in evaluating the acid-base balance of water. World Health Organization (WHO) has maximum acceptable limits of pH (6.5-8.5). The pH of the samples in the present study ranges (7.35-7.88) which is falling within the range of WHO limits. The overall results show that the Sirwan River water source is inside required and appropriate range.

3.2 Electrical Conductivity (EC), Total Dissolved Solids (TDS), and Turbidity

Unpolluted water is not a noble conductor of electric current. By increasing ion concentrations, electrical conductivity increases too [6]. Generally, (TDS) has the directly proportional with electrical conductivity. WHO has recommended standard tolerable limits for EC values that should not exceeded (400) μS/cm for drinking water [7]. The current study showed that the EC values were between (371-788) μS/cm. the results revealed that most of the results outside the WHO standard limits for drinking. However, for irrigation purpose the sample fall under medium category based on FAO standard.

Capacity of water is high to dissolve a numerous of inorganic and some organic minerals such as Ca$^{+2}$, K$^+$, Mg$^{+2}$, CO$_3^{-2}$, SO$_4^{-2}$, and NO$_3^{-}$. These dissolved ions formed
undesirable taste and color of water. Water with high TDS may affect persons suffering from heart and kidney diseases. WHO has a recommended maximum permissible limits of TDS which equals to (500) mg/L [8]. The concentration of (TDS) in samples was indicated in the range of (448-657) mg/L with an average of (515) mg/L. Therefore, it is out of the (WHO) standards limit for drinking purpose. The values of turbidity in the river are ranged between (1.47- 5.4) NTU, with the average value of (2.70) NTU. It can be considered as a safe limit [8]. Most of the samples do not exceed the turbidity limits (5 NTU). The permissible value for total hardness is (500) mg/l according to the WHO [8].

3.3 Dissolved oxygen (DO)

In water body, oxygen is obtainable in a dissolved state. DO is the concentration of oxygen that is dissolved in water [9]. DO is measured as one of the most vital characteristic of aquaculture. It is desirable by fish to breathe and perform metabolic activities. Lower levels of (DO) are frequently connected to fish kill happenings. The deficiency of DO may be owing to temperature, breathing, photosynthesis, aeration, and organic waste [9]. DO in current study water ranged from (6.7-7.99) mg/L, with the mean value of (7.51) mg/L. DO value with more than (5) mg/L is very important for fisheries life and production [10]. The results revealed that the DO is lower than desirable WHO limits in drinking water which equals to (8) mg/L.

3.4 Nitrate NO₃⁻

The allowable WHO maximum limits of nitrate is (45) mg/L in drinking water [11]. In current investigation, it is clear that the NO₃⁻ concentration ranges from (1.51-4.56) mg/L with the mean of (3.15) mg/L. These results show that amount of NO₃⁻ in the study sites are permissible.

3.5 Cations (Ca⁺², Mg⁺², Na⁺, and K⁺)

The concentrations of calcium were between (24.36-24.49) mg/l. All the samples were under acceptable and permissible limit. The concentrations of magnesium were between (17.95-20.17) mg/l. The water samples were within the permissible limits [12]. The samples analysis showed that the concentration of sodium vary between (12.35-17.83) mg/l. These were observed to be within permissible limit. For drinking water the acceptable limit for Na is about (200) mg/l [12]. On the other hand, potassium concentrations vary between (2.17-21.13) mg/l. Therefore, all samples had potassium concentration within the acceptable limit.
3.6 Anions (Cl\(^-\), SO\(_4^{2-}\), and NaHCO\(_3\)\(^-\))

The present study indicated that the Cl\(^-\) values were between (33.67-231.01) mg/L, and the mean is equals to 93.87 mg/L. the results were lower than WHO standard limits [12].

SO\(_4^{2-}\) is mainly obtained from the dissolution of salts sulfuric acid and almost in all water bodies found in abundance. Extraordinary concentration of SO\(_4^{2-}\) is may be because of oxidation of pyrite and excavation drain [13]. By now not most important negative influence of SO\(_4^{2-}\) on human health is informed. In the current investigation, SO\(_4^{2-}\) ion concentration was ranged from (135-165) mg/L with the average of (147) mg/L. The results revealed that the quantity of SO\(_4^{2-}\) in the study area is acceptable [11].

3.7 Carbonate and Bicarbonate:

They are existed in water because of some carbonate minerals present in water such as limestone, magnesite, and dolomite. This may influence pH values of water [14]. The concentrations of bicarbonate were between (258.64-378.2) mg/l with an average of (293.6) mg/l, or (3.98-5.81) meq/l with an average of (4.51) meq/l. So, all samples were found to be within moderate limits [12].

<table>
<thead>
<tr>
<th>Sample</th>
<th>Ca(^{2+})</th>
<th>Mg(^{2+})</th>
<th>Na(^+)</th>
<th>K(^+)</th>
<th>Cl(^-)</th>
<th>SO(_4^{2-})</th>
<th>HCO(_3^{-})</th>
</tr>
</thead>
<tbody>
<tr>
<td>Darbandixan</td>
<td>24.46</td>
<td>17.95</td>
<td>12.35</td>
<td>2.17</td>
<td>35.10</td>
<td>144.0</td>
<td>268.4</td>
</tr>
<tr>
<td>Maydan</td>
<td>24.49</td>
<td>18.49</td>
<td>12.39</td>
<td>2.17</td>
<td>119.70</td>
<td>136.0</td>
<td>258.6</td>
</tr>
<tr>
<td>Bawanur</td>
<td>24.45</td>
<td>19.58</td>
<td>15.92</td>
<td>2.27</td>
<td>33.67</td>
<td>156.0</td>
<td>341.6</td>
</tr>
<tr>
<td>Isayi</td>
<td>24.36</td>
<td>19.64</td>
<td>17.29</td>
<td>2.47</td>
<td>65.79</td>
<td>152.0</td>
<td>278.2</td>
</tr>
<tr>
<td>Qulasutyaw</td>
<td>24.45</td>
<td>19.37</td>
<td>15.59</td>
<td>2.35</td>
<td>43.29</td>
<td>135.0</td>
<td>378.2</td>
</tr>
<tr>
<td>Kalar</td>
<td>24.43</td>
<td>18.27</td>
<td>17.12</td>
<td>11.68</td>
<td>128.56</td>
<td>141.0</td>
<td>260.1</td>
</tr>
<tr>
<td>Shexlangar</td>
<td>24.45</td>
<td>20.17</td>
<td>17.83</td>
<td>21.13</td>
<td>231.01</td>
<td>165.0</td>
<td>270.0</td>
</tr>
<tr>
<td>Ave.</td>
<td><strong>24.44</strong></td>
<td><strong>19.07</strong></td>
<td><strong>15.50</strong></td>
<td><strong>6.32</strong></td>
<td><strong>93.87</strong></td>
<td><strong>147.0</strong></td>
<td><strong>293.6</strong></td>
</tr>
</tbody>
</table>

Concentrations are expressed in mg/L.

4. Water Quality Indexes

4.1 Sodium Adsorption Ratio SAR

It is a measure of the suitability of water for agricultural irrigation, as calculated from the ratio of Na\(^+\) to Ca\(^{2+}\) and Mg\(^{2+}\) by the following formula [15]:

\[
\text{SAR} = \frac{\text{Na}^+}{\sqrt{(\text{Ca}^{2+} + \text{Mg}^{2+}/2)}}
\]
Excess sodium in water leads to produce undesirable effects of changing soil properties and reducing soil permeability [16]. All the samples in the study area have SAR values within the excellent class and acceptable for irrigation.

4.2 Magnesium Hazard

It can be calculated through using the flowing equation that was proposed by (Szabolcs and Darb, 1964) [17]:

\[
MH (\text{meq/l}) = \frac{\text{Mg}^{2+}}{\text{(Ca}^{2+} + \text{Mg}^{2+})} \times 100
\]

The concentration of Mg$^{2+}$ ion can play an important role in soil productivity. When the value of magnesium hazard is less than (50), the water will be considered as safe and suitable for irrigation. The results of the water samples of the study area observed that all the samples have (MH) values greater than (50). Therefore, it cannot be used directly for irrigation without treatment or water management [17].

4.3 Kelly’s ratios KR

It is the concentration of Na$^+$ against Ca$^{2+}$ and Mg$^{2+}$ [18]. Water for irrigation uses was classified based on Kelly’s ratios. The Kelly’s ratio values less than (1) are considered suitable for irrigation [18].

4.4 Residual Sodium Carbonates RSC

It represents the amount of sodium carbonate and sodium bicarbonate in water when the total levels of carbonate and bicarbonate exceed the total amount of Ca$^{2+}$ and Mg$^{2+}$ [19]. Residual carbonate values with less than (1.25) are considered as safe. However, RSC values of (1.25-2.50) are within the marginal range. Those types of water should be used with good irrigation management techniques and soil salinity monitored by laboratory analysis [20]. RSC values of (2.50) or more are considered as high making the water unsuitable for irrigation use. RSC is determined through [19]:

\[
\text{RSC} = (\text{CO}_3^{2-} + \text{HCO}_3^-) - (\text{Ca}^{2+} + \text{Mg}^{2+})
\]

All ion concentrations are expressed in meq/l.
RSC values in the study area are ranges (1.2-3) with an average of (1.7). Therefore, most of the water samples are within the marginal range for irrigation except Qulasutaw station.

4.5 Exchangeable Sodium Percentage (ESP)

The desired value for ESP is (5) or less. However, values more than (5) mean increasing problems with soil infiltration and permeability, especially in clay soil. ESP value for irrigation water can be calculated from the following empirical relationship [20]:

\[ ESP = 100 \times \frac{-0.0126 + 0.01475 \times SAR}{1 + (-0.0126 + 0.01475 \times SAR)} \]

Therefore, most of the water samples are unsuitable for irrigation regarding ESP except tow stations (Darbandikhan and Maydan) which they have ESP values of less than (5).

Table 3: Water quality index results of the study area

<table>
<thead>
<tr>
<th>Sample</th>
<th>SAR</th>
<th>Mg%</th>
<th>KR</th>
<th>RSC</th>
<th>ESP</th>
</tr>
</thead>
<tbody>
<tr>
<td>Darbandixan</td>
<td>0.4</td>
<td>55.0</td>
<td>0.2</td>
<td>1.4</td>
<td>3.8</td>
</tr>
<tr>
<td>Maydan</td>
<td>0.4</td>
<td>55.7</td>
<td>0.2</td>
<td>1.2</td>
<td>3.8</td>
</tr>
<tr>
<td>Bawanur</td>
<td>0.5</td>
<td>57.2</td>
<td>0.2</td>
<td>2.4</td>
<td>5.5</td>
</tr>
<tr>
<td>Isayi</td>
<td>0.5</td>
<td>57.3</td>
<td>0.3</td>
<td>1.4</td>
<td>6.2</td>
</tr>
<tr>
<td>Qulasutyaw</td>
<td>0.5</td>
<td>56.9</td>
<td>0.2</td>
<td>3.0</td>
<td>5.4</td>
</tr>
<tr>
<td>Kalar</td>
<td>0.5</td>
<td>55.5</td>
<td>0.3</td>
<td>1.3</td>
<td>6.2</td>
</tr>
<tr>
<td>Shexlangar</td>
<td>0.5</td>
<td>57.9</td>
<td>0.3</td>
<td>1.2</td>
<td>6.4</td>
</tr>
<tr>
<td>Ave.</td>
<td>0.5</td>
<td>56.5</td>
<td>0.2</td>
<td>1.7</td>
<td>5.3</td>
</tr>
</tbody>
</table>

5 Heavy metals (total iron and Aluminum)

Although total iron levels low in natural water; however, it can be present in various ionic, organic and mineral forms. The concentration of total iron varied from (0.032-0.506) mg/L. Water samples at Bawanur, Qulasutaw, Kalar, and Shekhlanger showed high iron concentration than the prescribed limit by WHO which is (0.3) mg/L. On the other hand, the concentration of Aluminum in the analyzed samples was in the range of (0.093-2.411) mg/L which also higher than the prescribed limit by WHO. Aluminum levels in drinking water vary according to the levels found in the source water and whether aluminum coagulants are used during water treatment [21]. The sludge of the water treatment units are directly disposed into the river. This ultimately leads to increase the level of Aluminum and Iron in the river.

Therefore, water from the river cannot be used directly for drinking purpose. This is due to the nature of raw materials used in the industry and municipal wastes.
Table 4: The concentration of total Aluminum and Iron of study areas' samples

<table>
<thead>
<tr>
<th>Sample</th>
<th>T. Al</th>
<th>T. Fe</th>
</tr>
</thead>
<tbody>
<tr>
<td>Darbandixan</td>
<td>0.093</td>
<td>0.032</td>
</tr>
<tr>
<td>Maydan</td>
<td>0.491</td>
<td>0.149</td>
</tr>
<tr>
<td>Bawanur</td>
<td>2.411</td>
<td>0.445</td>
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<td>Isayi</td>
<td>0.423</td>
<td>0.086</td>
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<td>2.362</td>
<td>0.403</td>
</tr>
<tr>
<td>Kalar</td>
<td>0.975</td>
<td>0.506</td>
</tr>
<tr>
<td>Shexlangar</td>
<td>1.352</td>
<td>0.479</td>
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<tr>
<td>Ave.</td>
<td>1.158</td>
<td>0.300</td>
</tr>
<tr>
<td>WHO</td>
<td>0.200</td>
<td>0.300</td>
</tr>
</tbody>
</table>

6. Conclusion

On the basic of results, it was concluded that physicochemical properties revealed that most of the parameter such as pH, cations, and anions drop under the WHO permissible limits. However, there are some parameters like TDS, total Al, Fe fall outside of the permissible limits, and must be pretreatment before using for drinking. These parameters and. For irrigation purpose, data results show that the water of Sirwan River is medium salinity and may cause saline damages in the future. Also, some indexes including Mg hazards, ESP and RSC are with high values; therefore, water of the river should be used with good irrigation management techniques and soil salinity monitored by laboratory.

References

الخلاصة

تم إجراء هذه الدراسة لغرض تقييم جودة مياه نهر سيروان بين مجرى سد دربنديخان ومدينة كرميان لاستخدامات المنزلية والري. تم اختيار سبع محطات من مواقع مختلفة على طول نهر سيروان، وتم أخذ أربع نماذج مكررة من كل محطة. إن معايير جودة المياه المستخدمة في هذا البحث هي التكثيف، الدالة الهيدروجينية، الصرارة الكلية، الكالسيوم، الكربونات، الكربيد، التورسية الكهربائية، المواد الصلبة الذائبة الكلية. بين تحليل البيانات أن معايير جودة مياه نهر سيروان غير متوقفة مع معايير مياه الشرب خاصة تراكيز الالمنيوم والصوديوم الكلية. تظهر مستويات متزايدة من الحد المسموح به لمياه الشرب. بالإضافة إلى ذلك، وبناء على نتائج الدراسة، تظهر أن مياه نهر سيروان يجب أن تستخدم مع تقنيات إدارة جيدة للري ومتابعة نواتة للاراضي.

الكلمات المفتاحية: جودة المياه، نهر سيروان، تركيز الالمنيوم، نسبة امتصاص الصوديوم (SAR)، نسبة كاربونات الصوديوم المتبقية (RSC)، نسبة كاربونات الصوديوم المتبقية (RSC)، نسبة كاربونات الصوديوم المتبقية (RSC)، نسبة كاربونات الصوديوم المتبقية (RSC).